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# Topological Complexity in Protein Structures

Gabriella Heller  
*Pomona College*

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**2014**  
**Claremont Colleges Library**  
**Undergraduate Research Award**

**Senior Award**  
**Gabriella Heller**  
**Pomona College**

**Reflective Essay**

## Reflective Essay

My mathematics thesis project is perched at one of the many intersections of biochemistry and mathematics. In it, I apply concepts from topology to study the underlying structures of large biological molecules. Topologists are generally interested in whether one shape can be continuously deformed into another in space. To them, a coffee cup, doughnut, and CD are all equivalent because all three have a single hole and can be deformed from one to another without requiring tearing or gluing. Biochemists, on the other hand, are intrigued with the structure of proteins. Most traditional biochemists would chuckle at the notion of abstractly considering a highly ordered protein as a mere stretchy shoestring. In my thesis, however, I think about proteins as just that: deformable, shoestring-like mathematical objects that in rare cases contain knots, links, or other forms of topological complexity. Much to the surprise of classical biochemists, considering a protein in this way can yield valuable information about its stability and function.

It was my goal to present this thesis in such a way that it would be accessible to experts in both fields. The “bilingual” nature of my project called for a multiple-pronged approach to parsing through the literature across disciplines. The resources from the Claremont Colleges Library, such as the wide range of online database access, incredibly helpful library staff, and useful online guides, enabled me to seamlessly synthesize mathematics and biochemistry literature into a comprehensive final report that integrates perspectives across these disparate fields and offers new insights.

Firstly, my project would not have been possible without access to the online resources made available to me by the library. The mathematics component of my research relies heavily on MathSciNet, a database created by the American Mathematical Society. Access to the articles posted on MathSciNet requires a hefty single subscriber rate for individuals outside of a consortium of over \$12,000. Without the library’s support, this resource would simply be inaccessible to me. A second essential database to my project is SciFinder, a database that gives me unparalleled access to references published in the sciences. Just as in the case of MathSciNet, SciFinder would be inaccessible to me without the library’s support as a result of its \$3,000 individual subscription cost.

MathSciNet and SciFinder not only assisted me in finding the relevant resources for my topic and directed me to the citations of a given reference, but also showed me which more recent works have cited a given article. This enabled me to access the most recent literature about the underlying topological structures of proteins. While many of these more recent articles contained crucial information for my project, I was unable to find them using simple search methods such as Google Scholar, likely due to their relatively low citation numbers. This new information not only ensured that my thesis includes the most relevant, up to date information, but also helped shape the direction of my project. For example, using SciFinder, I came across a paper that I had previously not seen entitled “Protein knots and fold complexity: some new twists” by William Taylor. In this paper, Taylor argues why one specific knot is not observed within protein structures, while another similar one is observed. In my thesis, we use his argument as the basis for a section on hypothetical protein structures to propose how a protein could obtain this missing knot type.

It was not only the library’s online database access that assisted me with this research project, but also my very positive interactions with the library’s staff. In fact, I only became introduced to SciFinder as a result a workshop presented by Librarian Sean Stone to my Organic Chemistry class. During that workshop, I learned the power of the library’s tools offered us when making complex queries to the previously imposing scientific literature base. Were it not for the helpful step-by-step guidance of Mr. Stone during that workshop I am sure that I would be less capable of searching through the enormous body of relevant literature.

Another extremely helpful service that the library offers is the online chat feature. The majority of my mathematics thesis was completed over winter break while I was off campus. Although I initially struggled obtaining off-campus access to the library’s resources, I soon realized that the library’s assistance was at my fingertips with the “Ask Us” chat feature built right into the library’s website. An anonymous staff member quickly assisted me with off campus access, giving me access to all the invaluable articles that I would normally have access to while on campus.

Besides the many ways the library staff helped me one-on-one, their informative online guides provided crucial information even when in-person help wasn’t available. Just as chemists and mathematicians have different ways of thinking about three dimensional objects, they also

have different ways of presenting their results. Chemists typically use the straightforward and practical Microsoft Word to format their documents and EndNote for citation purposes, whereas mathematicians generally use the involved, yet stunning LaTeX and BibTeX for formatting and citing, respectively. As this was a mathematics thesis, I opted to use LaTeX and BibTeX, despite being relatively unfamiliar with how to cite my references in this new format. Luckily for me, Librarian Sam Kome created an extremely useful online guide entitled “Getting BibTeX Citations,” which greatly simplified this processes by describing how to import citations directly into the BibTeX citation manager.

The dozens of citations in my thesis echo the library’s crucial role in its creation. Beyond guiding me to these references, the library shaped the direction of my thesis and assisted me in its final presentation. The library’s resources, friendly staff, and wonderful informative online guides have enabled me to produce a work of which I can be proud.

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Gabriella Heller  
Pomona College

Senior Thesis

“Topological Complexity in Protein Structures”

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2014

# Topological Complexity in Protein Structures

Gabriella Heller  
*Pomona College*

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GABRIELLA HELLER

TOPOLOGICAL  
COMPLEXITY IN  
PROTEIN STRUCTURES

ADVISOR: PROFESSOR FLAPAN  
POMONA COLLEGE MATHEMATICS DEPARTMENT



Submitted in partial fulfillment of the requirements  
for the award of Bachelors of Arts in Mathematics.

*Dedicated to Erica Flapan, Ami Radunskaya, and Matthew McDermott. Thank you for inspiring me to study mathematics.*

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## Abstract

For DNA molecules, topological complexity occurs exclusively as the result of knotting or linking of the polynucleotide backbone. By contrast, while a few knots and links have been found within the polypeptide backbones of some protein structures, non-planarity can also result from the connectivity between a polypeptide chain and attached metal structures. In this thesis, we survey the known types of knots, links, and non-planar graphs in protein structures with and without including such bonds between proteins and metals. Then we present new examples of protein structures containing Möbius ladders and other non-planar graphs as a result of these bound metal atoms. Finally, we propose hypothetical structures illustrating specific disulfide connectivities that would result in the key ring link, the Whitehead link and the  $5_1$  knot, the latter two of which have thus far not been identified within protein structures.



## CHAPTER 1

# Introduction

Until two decades ago, the underlying form of all native protein structures was assumed to be topologically linear. This was in contrast to DNA molecules whose polynucleotide backbones had been known for a decade to exist in both linear and closed circular knotted and linked forms [4, 5, 35]. In 1994 and 1995, Liang and Mislow [8, 9, 10] showed that if cofactors (non-protein chemical compounds) and disulfide bonds (strong bonds between nonadjacent regions of a protein) are taken into account when evaluating the topological form of protein structures, then these structures may indeed contain knots, links, and even non-planar graphs. More recently, even when considering only the polypeptide backbones of proteins, some knots and links have been identified [3, 12, 15, 23, 30, 31, 34].

Why some proteins contain these non-planar features has been a question since their discovery. The fact that features such as knots and non-planar spatial graphs have been highly preserved throughout evolution suggests that they may play a crucial role in enzymatic activity. For example, in one protein, ubiquitin hydrolase, the 5 fold knot is highly conserved from human to yeast forms, thus suggesting its importance [27]. In cases when a protein loses or gains a knot through the process of evolution, a change in function of the protein is observed. Furthermore, proteins containing knots and links have been observed to have increased stability [34].

We are interested in studying the topological complexity of the underlying structures of proteins. We define an object to be *topologically complex* if it is non-planar in the sense that even assuming complete flexibility and elasticity, it cannot be deformed into a plane. For example, a knotted circle is non-planar and hence is indeed a type of topological complexity. Understanding how such non-planarity arises in proteins may offer valuable insight into protein stability, folding mechanisms, and degradation pathways.

In order to evaluate their topological complexity, we model protein structures as completely flexible objects. While a protein molecule is in fact only partially flexible about its peptide bonds, if we can show that a completely flexible model cannot be deformed into a plane, then the protein, with less flexibility than the model, also cannot attain a planar conformation. Thus, for the purpose of identifying non-planarity within protein structures, we can assume complete flexibility, even though this is not physically accurate. Figure 1 illustrates how, for a given protein with a structured geometric shape, we associate a flexible model that we refer to as its *underlying topological structure*. Because the backbone of the protein in the

figure contains neither disulfide bonds nor other cofactors connecting non-adjacent regions, its underlying topological structure is an open linear path.

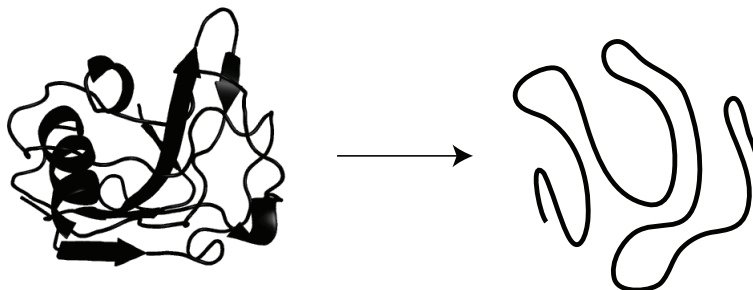


FIGURE 1. We represent the protein ribbon diagram on the left by a flexible underlying topological structure, shown on the right.

In Chapter 2, we present a survey of non-planar topological structures that have been previously identified within proteins. Then in Chapter 3, we present new examples of proteins containing the non-planar graphs  $K_{3,3}$  and  $K_5$  as well as some containing Möbius ladders as a result of covalently-bound cofactors. Finally, in Chapter 4, we propose hypothetical protein structures whose covalent bonding would result in a 3-component key ring link, the Whitehead link, and the  $5_1$  knot, the latter two of which have not been previously identified in the underlying structures of proteins.



## CHAPTER 2

# Survey of Known Non-Planar Protein Forms

### 1. Knots

Mathematicians define a *knot* as a circular path in three dimensional space. See, for example, the image on the left side of Figure 2. The requirement that the path be circular is so that the knot is topologically trapped within the figure. By contrast, a knot which is contained in an open linear segment (as shown on the right in Figure 2) can be removed by unthreading.

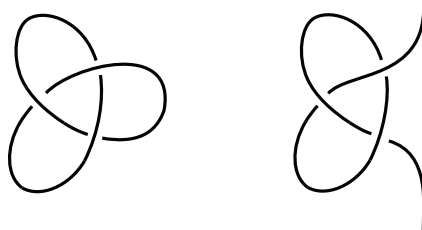


FIGURE 2. The knot on the left is trapped in a circular path, whereas the knot on the right can be unknotted by a deformation.

While the polypeptide backbones of protein structures are normally open segments rather than circular paths, if a knot is deeply embedded in a protein structure, then from a biochemical viewpoint, it is reasonable to assume it is trapped within the structure. This assumption generally holds as a result of the knot's location in a valley of the potential energy landscape. In particular, the energy required for the termini, usually near the protein surface, to unthread the knot is so prohibitively large that this event generally does not occur. However, it is a difficult problem to create a model of such a knotted open segment in which the knot is topologically trapped within the model and can be uniquely identified.

Many approaches have been developed to creating such a topological model [1, 6, 13, 22, 28, 33]. One such approach is to extend the termini indefinitely in opposite directions [7]. We illustrate this by adding arrows at the ends of the linear segment as shown in Figure 3. However, if one or both of the termini are near the tangling of the knot, then different knots can be trapped in the structure depending on how the ends are extended [17, 19, 25]. For example, in the right image of Figure 3, if we extend the bottom endpoint under the arc we will get a trefoil knot, but if we extend it over the arc we will get the unknot.

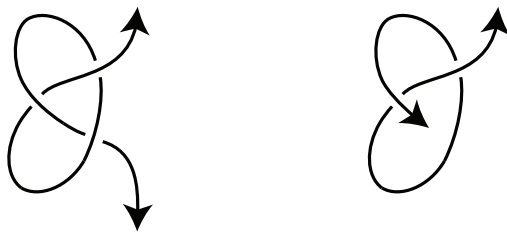


FIGURE 3. For the image on the left we can extend the endpoints of the segment indefinitely so that the knot is trapped and is uniquely determined. But for the image on the right, we may obtain either a knot or an unknot depending on how we extend the bottom endpoint.

Another approach to modeling a knotted linear protein structure is to place it in the center of a large sphere. The termini are then extended to the spherical boundary of the ball and joined by an arc on that boundary to create a closed loop (see Figure 4) [13, 28]. However, the same problem can occur that we saw on the right side of Figure 3 when we extended the ends indefinitely. In particular, different knots may be obtained depending on how the arcs are extended to the boundary of the ball. To resolve this problem, Millett and others [11, 16, 18, 20, 21] have proposed a statistical approach in which they assign different probabilities to each of the possible knots they obtain in this way. Then, only the knot with the highest probability is considered to occur in the given protein structure.

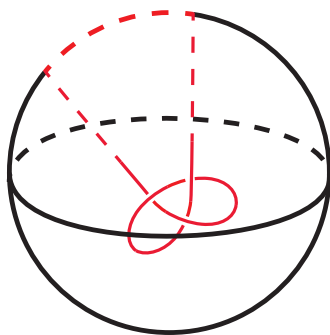


FIGURE 4. One way to determine whether a protein contains a knot is by placing the protein in a large sphere, extending the termini to the spherical boundary and joining them by a closed arc along the surface of the sphere. Here, this concept is illustrated by a red trefoil-like, unknotted structure (solid red line) that is placed inside a sphere. By extending its termini to the boundary of the sphere, and joining the ends by an arc along the sphere's surface (red dashed lines) we see that we can view this structure as a knot.

Whatever method is taken to modeling and characterizing knots in proteins, amongst the tens of thousands of known structures, only a few hundred have been

found to contain knots. Furthermore, every one of the known knotted proteins is deformable to one of the knots  $3_1$ ,  $4_1$ ,  $5_2$  and  $6_1$ , illustrated in their circular forms in Figure 5.

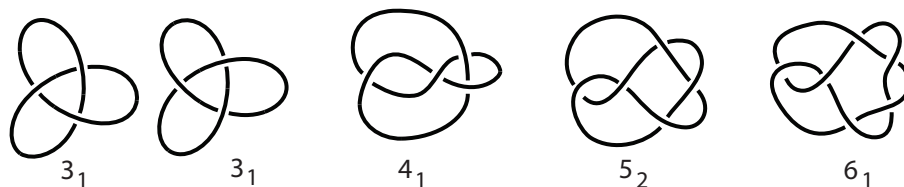


FIGURE 5. These are the only knot types that have been identified in protein structures to date.

Topologists refer to each knot by a number together with a subscript. The first number indicates the minimum number of crossings required in any illustration of the knot. Since this number does not necessarily uniquely determine the knot, the subscript is then used to distinguish knots with the same minimum number of crossings. While the trefoil knot  $3_1$  and the figure eight knot  $4_1$  are the only knots containing three and four crossings respectively, there are two distinct knots  $5_1$  and  $5_2$  containing five crossings and three distinct knots containing six crossings. For knots which are topologically distinct from their mirror image, the same pair of numbers is used to refer to both the knot and its mirror image. Of the knots illustrated in Figure 5, the figure-of-eight knot  $4_1$  is the only one which is topologically equivalent to its mirror image. Figure 5 illustrates both mirror forms of the  $3_1$  knot, since both have been identified in proteins. The  $5_2$  and  $6_1$  knots have each only been found in proteins in the form illustrated in the Figure (see [24]).

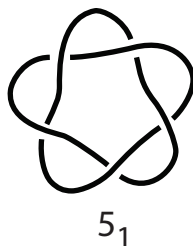


FIGURE 6. This five crossing knot that has yet to be found in a protein structure.

It is perhaps surprising that the relatively simple knot  $5_1$  knot (illustrated in Figure 6) has yet to be found in any protein structure. Taylor has hypothesized that this knot is an unlikely motif in protein structures, because a rate limiting “double threading” event would be required for its formation [32]. Taylor contrasts this with the  $5_2$  knot, which could be the result of threading one end of the backbone through a twisted hairpin, and then connecting the termini with an arc to create a closed loop as in the model introduced by Millet and others (discussed above). We

illustrate this in Figure 7. In Chapter 3, Section 3, we discuss how the  $5_1$  knot may hypothetically occur in a protein structure with and without considering disulfide bonds.

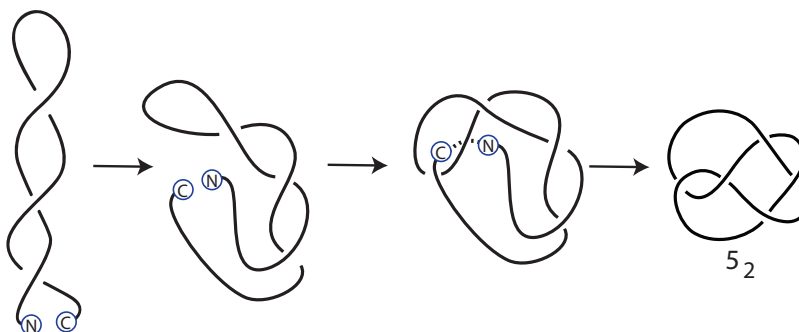


FIGURE 7. The  $5_2$  knot could be the result of threading one end of the backbone through a twisted hairpin as illustrated.

The first knotted protein structures to be identified were metalloproteins ascorbate oxidase and human lactoferrin, which were found by Liang and Mislow to contain the trefoil knot  $3_1$  [8]. The existence of these knots relied on the covalently bound metal atoms and disulfide bonds of the protein. For example, we can see in Figure 8 that the trefoil knot found in ascorbate oxidase makes use of the thick grey segments representing disulfide bonds.

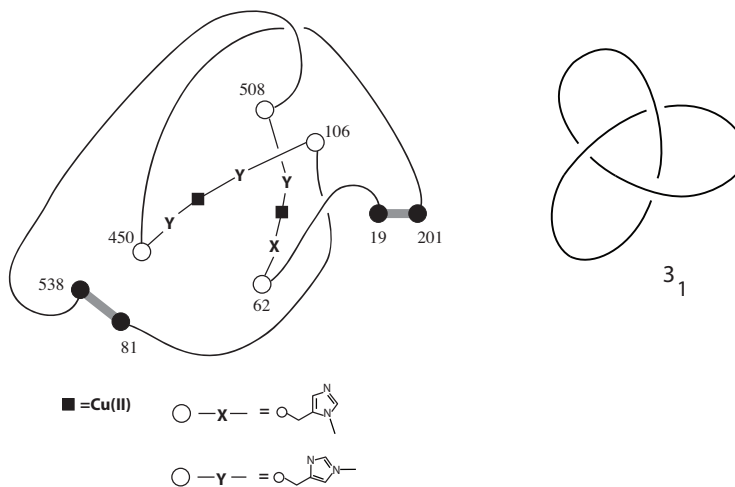


FIGURE 8. The trefoil knot contained in ascorbate oxidase makes use of the thick grey segments representing disulfide bonds.

More recently, nearly forty proteins have been identified which have knots entirely formed by their polypeptide backbones. The trefoil knot is by far the most

common knotted motif reported in protein backbones, and is known to exist in proteins in both the left- and right-handed forms [31]. By contrast, only a few proteins are currently known to contain the  $4_1$  knot, and only individual examples of the  $5_2$  and  $6_1$  knots, respectively, are known [34].

In 2006, the first knot with five crossings was identified in the backbone of a protein by Virnau et. al. This  $5_2$  knot is contained in the structure of ubiquitin hydrolase, a protein responsible for rescuing proteins from degradation. It is believed that the complex knotted structure of this protein helps protect it against degradation. This theory is supported by the fact that protein degradation is generally initiated by an event in which proteins are threaded through a narrow pore to remove their ordered structure. It is plausible that the steric bulk from a knot may block the protein from being threaded through such a pore, thus potentially preventing the ultimate degradation of the molecule. Further experiments must be carried out to validate this theory. [34]. The most complex protein knot that has been found to date is a Stevedore's knot  $6_1$ , which was identified in 2010 [2].

Table 1 gives a list of the knots, links, and non-planar graphs found in proteins to date and a brief description of how they arise (e.g., from the backbone, from cofactors, etc.). We explain more about the links and spatial graphs in this table in later chapters of this thesis. Note that the table provides a summary of the types of topological diversity that have been identified in protein structures, but does not provide a comprehensive list of all proteins that contain non-planar features. More extensive lists of knotted proteins can be found elsewhere [24, 31, 34, 37].

## 2. Links

We now define a *link* as two or more disjoint circular paths in three dimensional space which cannot be separated from each other. Here, by “separated from each other” we mean that there is no deformation of the link so that the two rings lie on opposite sides of a plane in  $R^3$ . For example, the pair of circles on the left in Figure 9 is linked, whereas the pair of circles on the right is unlinked because the pair can be deformed so that one circle is on either side of a vertical plane. The most common type of protein linking has the form of the *Hopf link*, which consists of two circles linked together only once as illustrated on the left of Figure 9.

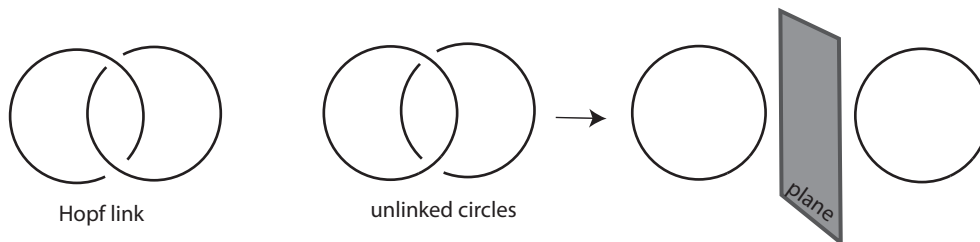


FIGURE 9. The pair of circles on the left is a Hopf link, but the pair of circles on the right is unlinked.

TABLE 1. Selected Examples of Topological Complexity Previously Found in Protein Structures

Knots		
$3_1$	<b>cofactors used:</b> ascorbate oxidase, human lactoferrin <b>backbone only:</b> hypothetical protein, plasmid pTiC58 VirC2, N-succinyl-L-ornithine transcarbamylase, methyltransferase domain of human TAR RNA binding protein, alpha subunit of human S-adenosyl-methionine synthetase, human carbonic anhydrase II, ribosomal 80S-eEF2-sordarin complex	[8, 10, 37, 24, 30, 34]
$4_1$	<b>backbone only:</b> acetohydroxy acid isomeroreductase, photosensory core domain of aeruginosa bacteriophytochrome	[31, 37, 24]
$5_2$	<b>backbone only:</b> ubiquitin hydrolase	[24, 37]
$6_1$	<b>backbone only:</b> $\alpha$ -haloacid dehalogenase	[24]
Links		
Hopf	dimeric citrate synthase from <i>P. aerophilum</i> (two subunits are linked by two intramolecular disulfide bonds), bovine mitochondrial peroxiredoxin III (two interlinked rings, each consisting of twelve subunits), tuna cytochrome c (the porphyrin makes up one ring and the protein backbone together with central heme iron atom make up another)	[3, 37, 10]
Key Ring	Cytochrome c3 from <i>Desulfovibrio vulgaris</i> Miyazaki	[10]
Chain Mail	bacteriophage, HK97 (72-component linked chain mail)	[36, 37]
Non-Planar Graphs		
$K_{3,3}$	Fe <sub>4</sub> S <sub>4</sub> cluster of chromatium high potential iron protein	[9]
$K_5$	Fe <sub>4</sub> S <sub>4</sub> cluster of chromatium high potential iron protein	[9]

The Hopf link has been found in protein structures in several distinct ways including the linking of two polypeptide chains via intramolecular disulfide bonds, the linking of multi-subunit macromolecular non-covalent rings [3, 37], and linking resulting from the inclusion of porphyrin cofactors in tuna cytochrome c [10]. These three different forms of protein Hopf links are illustrated in Figure 10.

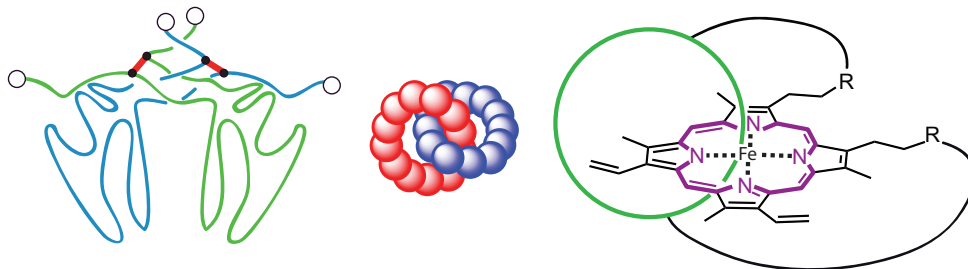


FIGURE 10. Examples of Hopf links in protein structures. Left: dimeric citrate synthase from *P. aerophilum* in which two subunits are linked as a result of intramolecular disulfide bonds, shown in red. Center: bovine mitochondrial peroxiredoxin III in which two interlinked rings, each consisting of twelve subunits, form the Hopf link. Right: tuna cytochrome c in which the porphyrin makes up one ring (purple) and the protein backbone together with central heme iron atom make up another (green).

One way to describe the linking between two closed curves in three-dimensional space is with *linking number*. Intuitively, the linking number represents how many times one curve winds around the other. Before we can compute the linking number however, we must first introduce *oriented links*, or links with an arrow assigned to each component to indicate a particular direction. When a link has oriented components, its crossings can be identified as negative or positive crossings. A positive crossing corresponds to a right handed twist and a negative crossing corresponds to a left handed twist, as shown in Figure 11.

Thus, we can define *linking number* as follows: Let  $L_1$  and  $L_2$  be components of an oriented link projection  $L$ . The *linking number* of  $L$  is one half the number of positive crossings between  $L_1$  and  $L_2$  minus the number of negative crossings between  $L_1$  and  $L_2$ . Interestingly, the links presented in Figure 10 all have linking number  $\pm 1$ . This is what defines a Hopf Link.

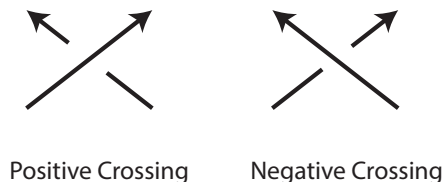


FIGURE 11. A positive crossing and a negative crossing.

Other types of links can be found as a result of multiple heme cofactor binding. Heme is a small chemical compound that takes the form of an organic ring with an iron atom in its center. For our purposes, we can visualize it as a simple ring, as shown in purple in Figure 12. In particular, by treating each of a protein's heme cofactors as its own ring, cytochrome  $c_3$  can be seen to contain a five-component link [8]. By omitting the pink arcs in Figure 12, we obtain the key ring link illustrated on the right. The linking number between a purple and green component will be  $\pm 1$ , whereas the linking number between any two purple components will be 0.

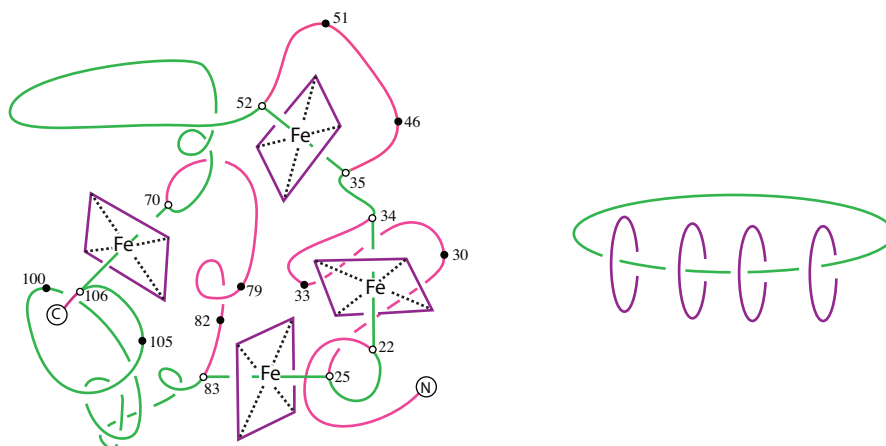


FIGURE 12. Cytochrome  $c_3$  from *Desulfovibrio vulgarism* Miyazaki and its underlying five-component key ring link.

Perhaps the most fascinating example of linking in protein structures is the linking that exists to protect the bacteriophage, HK97. On the exterior of the bacteriophage, 72 rings link together to form a chainmail protective capsid, illustrated in Figure 13. This structure is made up of 12 pentagonal rings and 60 hexagonal rings linked together to form a spherical shape with icosahedral symmetry, meaning that it has 60 rotational symmetries. Each pentagonal component is linked to five other components such that each link formed has linking number  $\pm 1$ . Similarly, each hexagonal component is linked to six other components, such that each link formed has linking number  $\pm 1$ . These components come together to create a protective surface that is relatively thin, and it is believed that the topological linking adds stability to the capsid [36, 37].



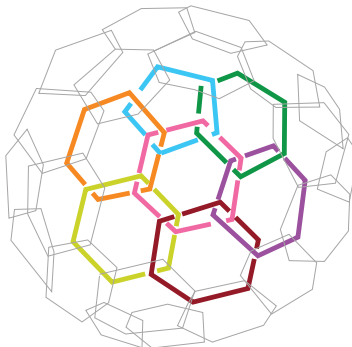


FIGURE 13. A chain mail protein link made up of 12 pentagonal rings and 60 hexagonal rings in a spherical shape with icosahedral symmetry.

Whether a protein link contains two, five, or 72 components, linking seems to greatly increase stability of the resulting structure, likely by reducing the entropy of the unfolded state [37]. As more examples of topological links are discovered in the underlying structures of proteins, it will be fascinating to study the mechanism of their formation.

### 3. Spatial Graphs

As we saw above, protein structures can be non-planar because they contain a knot or link. However, these are not the only types of topological non-planarity that can exist within proteins. In particular, non-planar graphs have been found in certain proteins that contain metal clusters. In order to determine whether such a structure contains a non-planar graph, we represent the structure as a graph in three dimensional space where vertices represent atoms or groups of atoms and edges represent bonds. For example, Figure 14 illustrates a spatial graph representing the  $\text{Fe}_4\text{S}_4$  cluster.

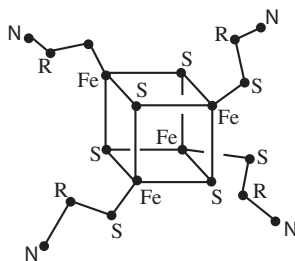


FIGURE 14. A spatial graph representing the  $\text{Fe}_4\text{S}_4$  cluster.

We begin by introducing the following two important families of graphs.

DEFINITION. A complete graph on  $n$  vertices, denoted by  $K_n$ , is a graph in which every pair of vertices is connected by an edge.

We are particularly interested in the complete graph on five vertices  $K_5$  (illustrated in Figure 15) because it is the smallest complete graph which cannot lie entirely in a plane.

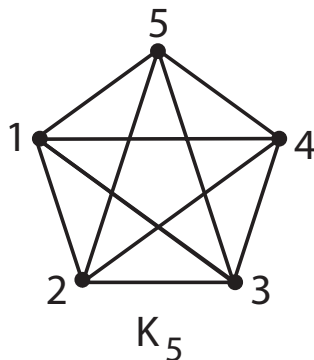


FIGURE 15. The complete graph on five vertices  $K_5$ .

DEFINITION. A complete bipartite graph  $K_{m,n}$  on two sets of vertices, one containing  $m$  vertices and the other containing  $n$  vertices, is a graph in which every vertex in the set of  $m$  vertices is connected to every vertex in the set of  $n$  vertices, but there are no edges connecting a pair of vertices both in the same set of vertices.

We are particularly interested in the complete bipartite graph  $K_{3,3}$  (illustrated in Figure 16) because it is the smallest complete bipartite graph which cannot lie entirely in a plane.

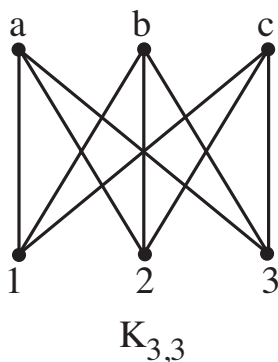


FIGURE 16. The complete bipartite graph  $K_{3,3}$

Observe that the illustrations in Figures 15 and 16 appear as though some edges intersect, which is not the case. We have drawn the graphs in this way because

we are focusing on the connectivity between vertices, rather than on a particular conformation of the graph in space. When we talk about a graph exclusively in terms of its connectivity, we refer to it as an *abstract graph*, whereas when we talk about a conformation of a graph in space we refer to the graph as a *spatial graph*. Thus Figures 15 and 16 represent the abstract graphs  $K_5$  and  $K_{3,3}$ .

If a non-planar graph is contained in a protein structure, it will prevent the protein from being deformed into a plane. However, a non-planar graph can play this role not only if it is a subset of the underlying structure of the protein, but also if it can be obtained from the underlying structure by collapsing edges and/or omitting edges or vertices. If the non-planar graph is obtained from a structure in this way we say it is a *minor* of the structure. For example, in Figure 17, the graph on the right is obtained from the graph on the left by collapsing the red edges and deleting the blue vertices and blue edge. Thus the graph on the right is a minor of the graph on the left.

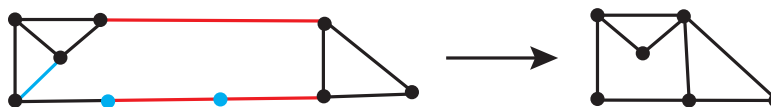


FIGURE 17. The graph on the right can be obtained by collapsing the red edges and deleting the blue vertices and blue edge. Hence the graph on the right is a minor of the one on the left.

Mislow and Liang used the following well-known theorem to show that certain protein structures containing cofactors are topologically non-planar [9].

**KURATOWSKI'S THEOREM.** *A graph can lie in a plane if and only if it does not contain either of the graphs  $K_5$  or  $K_{3,3}$  as a minor.*

One of Mislow and Liang's examples is *Chromatium* high potential iron protein, which contains the  $\text{Fe}_4\text{S}_4$  cluster [9]. By Kuratowski's Theorem, showing that the structure contains either a  $K_5$  or  $K_{3,3}$  graph as a minor is sufficient to show the structure is non-planar. The incorporation of the cluster into the backbone ensures, in fact, that it contains both of these graphs as we explain below.

The left image in Figure 18 illustrates the underlying graph of *Chromatium* high potential iron protein. In order to show that it contains  $K_{3,3}$  as a minor, we need to identify two sets of three vertices such that every vertex in one set is connected by a disjoint path to every vertex in the other set. One of these sets of vertices (shown in green in the central image) will consist of three of the four iron atoms that occur on the corners of the cube-like metal structure. The other set of vertices (shown in pink) will consist of two sulfur atoms on the corners of the cube-like structure together with one sulfur atom from a cysteine residue on the polypeptide backbone. The orange, purple, and black segments form disjoint paths from every green vertex to every pink vertex. The blue vertices are deleted from the middle graph to obtain

the  $K_{3,3}$  graph. We illustrate  $K_{3,3}$  as an abstract graph on the right with the same color coding to demonstrate that the sets of green and pink vertices together with the orange, purple and black edges do indeed form a  $K_{3,3}$ . It now follows from Kuratowski's theorem that this protein structure is non-planar.

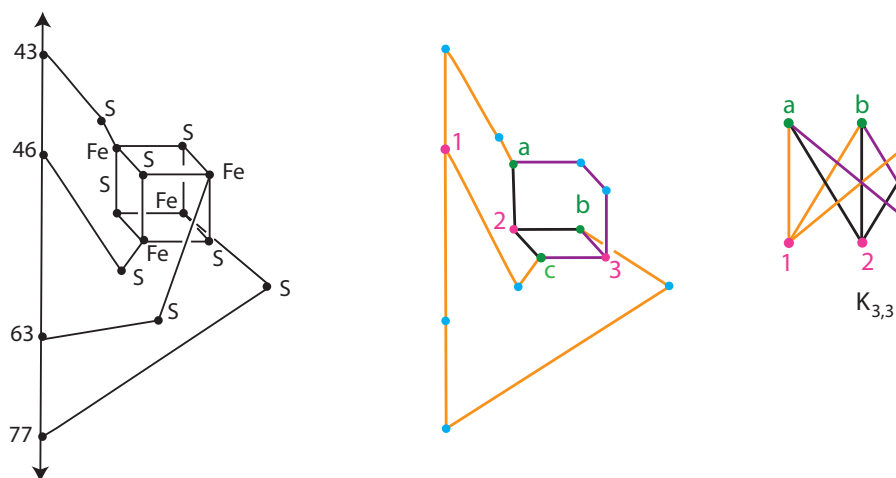


FIGURE 18. The left graph illustrates the cluster connected to the polypeptide backbone. The numbers in the left graph identify amino acid connectivity to the backbone. In the middle graph, six vertices have been partitioned into a pink group and a green group such that there is a path from every green vertex to every pink vertex. Thus the structure contains a  $K_{3,3}$  graph, and hence is non-planar.

In Figure 19, we illustrate that the underlying graph of *Chromatium* high potential iron protein also contains  $K_5$  as a minor as a result of the same  $\text{Fe}_4\text{S}_4$  cluster. In particular, starting with the graph on the left, we collapse the red edges and delete the blue vertices and blue edges to obtain the graph on the right which has five vertices numbered 1 through 5 and disjoint paths between every pair of vertices. Thus, this cluster also contains  $K_5$  as a minor. This fact alone is also sufficient to show *Chromatium* high potential iron protein is non-planar.

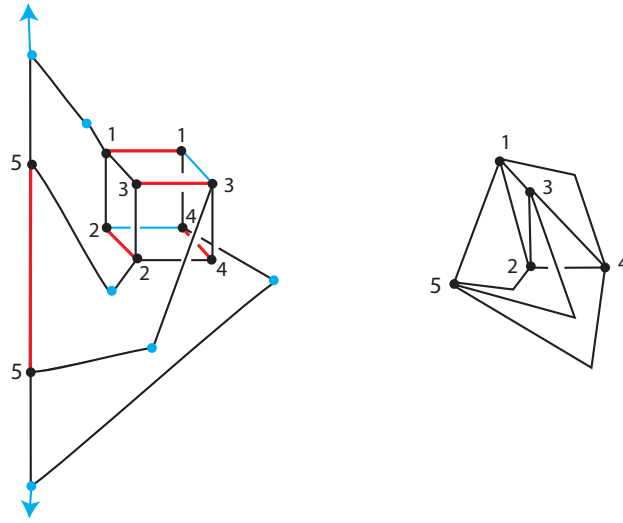


FIGURE 19. The left graph illustrates the cluster as it is connected to the polypeptide backbone. We collapse the red edges and delete the blue vertices and edges to obtain the  $K_5$  graph on the right.



## CHAPTER 3

# Our Results

### 1. Non-planarity in Nitrogenase

Nitrogenase is a protein complex that plays a crucial role in the global nitrogen circulation, catalyzing the reduction of atmospheric nitrogen into biologically available ammonia. This complex consists of two proteins. The first, often referred to as the Fe protein contains two subunits connected by an  $\text{Fe}_4\text{S}_4$  cluster. The second protein, often referred to as the MoFe protein, contains a total of four complex metal clusters: two P-clusters  $[\text{Fe}_8\text{S}_7]$ , shown on the right in Figure 20, which occur at subunit interfaces and two M-clusters  $[\text{MoFe}_7\text{S}_9\text{C-homocitrate}]$ , shown on the left in Figure 20, which are embedded in the protein. Understanding the structure of these P- and M- clusters had challenged biochemists for decades. In fact, the true elemental composition and connectivity was only established in 2011 [29]. We will show below that as a result of these cofactors, nitrogenase contains several topologically non-planar structures.

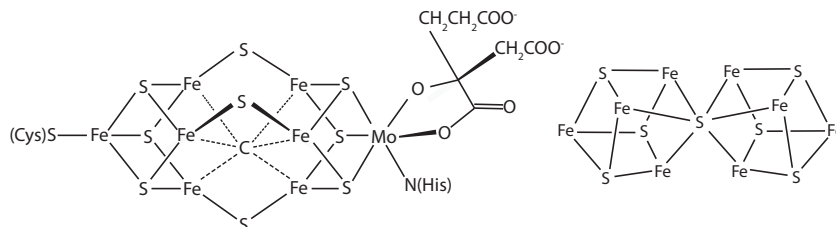


FIGURE 20. The M- and P- clusters found in nitrogenase shown on the left and right, respectively. Thick wedges represent bonds that are coming out of the page, dashed lines illustrate that the carbon atom (C) is in the center of the cluster.

The first metal cluster that we consider is the M-cluster. In order to use Kuratowski's Theorem to prove the non-planarity of nitrogenase, we show that the M-cluster contains a  $K_{3,3}$  graph. In particular, in Figure 21, we define one set of three vertices by labeling the top left sulfur atom, the central carbon, and the top right sulfur atom by the letters  $a$ ,  $b$  and  $c$ , respectively. We define the second set of three vertices by labeling the top left iron, bottom left iron, and central right iron atoms vertices by the numbers 1, 2, and 3, respectively. We use purple, green, and orange to highlight disjoint paths from vertices  $a$ ,  $b$ , and  $c$  to every numerical vertex.

Thus the M-cluster contains a  $K_{3,3}$  graph and hence by Kuratowski's theorem, is indeed non-planar.

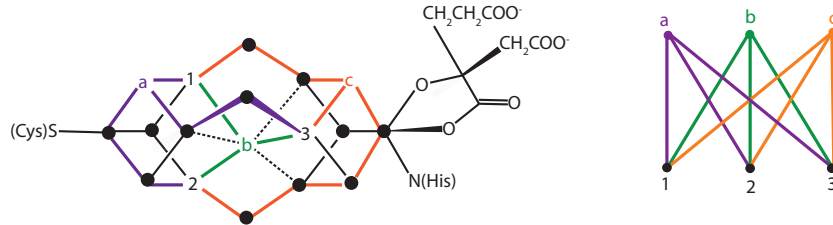


FIGURE 21. The M-cluster of nitrogenase contains a  $K_{3,3}$  graph, and hence is non-planar.

We see as follows that the P-cluster together with the two subunit backbones of nitrogenase contains both of the non-planar graphs  $K_{3,3}$  and  $K_5$  as minors. Figure 22 illustrates two different  $K_{3,3}$  graphs contained in this structure. Notice that these graphs each make use of part of the pink backbone connected to the cluster. In fact, the P-cluster together with the two backbones contains additional  $K_{3,3}$  graphs (not illustrated here).

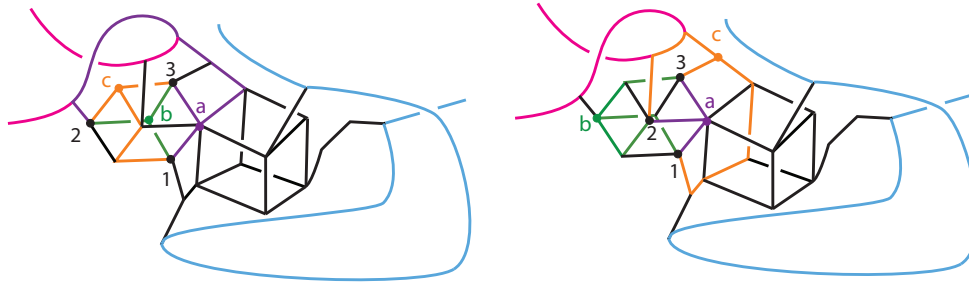


FIGURE 22. The P-cluster together with the pink backbone contains multiple  $K_{3,3}$  graphs.

In order to see that the graph  $K_5$  is also a minor of nitrogenase, we begin with the colored image on the left in Figure 23. Then we collapse the red edge and delete the blue vertices, blue edges, and the green and pink arcs of the backbones. Thus we obtain the purple  $K_5$  illustrated on the right as a minor of the graph on the left. Observe that this  $K_5$  makes use of an arc on the pink backbone in order to join vertex 4 to vertex 5 and an arc on the green backbone in order to join vertex 2 to vertex 5.



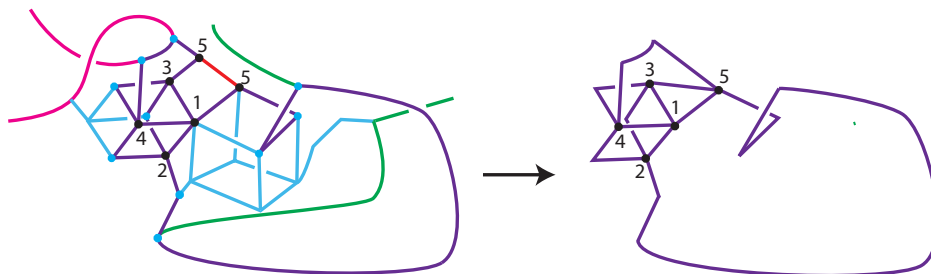


FIGURE 23. We collapse the red edge and delete the blue vertices, blue edges, and the green and pink arcs of the backbones to obtain  $K_5$  as a minor of the structure on the left.

## 2. Möbius Ladders

Just as a circular path may contain different knots depending on its conformation in space, an abstract graph can have topologically distinct spatial conformations. One particular conformation of the abstract graph  $K_{3,3}$  is known as a *Möbius ladder* with three rungs. Möbius ladders are noteworthy because they can resemble Möbius strips.

A Möbius strip is defined as a one-sided nonorientable surface that is obtained by cutting a closed band into a strip, giving one of the two ends a half-twist and reattaching the two ends. A Möbius ladder can be defined analogously to a Möbius strip, except instead of starting with a closed band, we begin with two circles connected by rungs. We illustrate that the graph  $K_{3,3}$  is equivalent to a Möbius strip with three rungs in Figure 24. While the figure on the left looks quite different from the one on the right, it is easy to check that as abstract graphs the two figures have exactly the same connectivity.

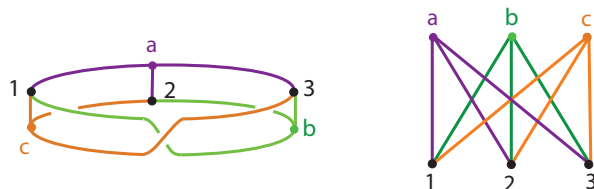


FIGURE 24. A conformation of  $K_{3,3}$  which has the form of a Möbius strip is known as a *Möbius ladder* with three rungs.

We saw in Figure 21 that the M-cluster of nitrogenase contains the abstract graph  $K_{3,3}$ . Figure 25 shows that, in fact, the spatial conformation of  $K_{3,3}$  in the M-cluster from Figure 21 can actually be deformed to a Möbius ladder with three rungs. We explain the steps of this deformation as follows. The first step pulls vertex 3 down and towards the right in front of the edge connecting vertices 2 and  $c$ . The second step rounds out the circular path that goes from vertex 1 to vertex  $b$

to vertex 3 to vertex  $c$  to vertex 2 to vertex  $a$  and then back to vertex 1 so that it looks like a sideways eight. In the third step, the edge connecting vertices  $a$  and 3 is pulled down below the rest of the figure. The fourth step rotates the entire graph by  $90^\circ$  so that the eight is now vertical. In the fifth step, vertex 2 is slid under the crossing and up to the left. In the last step the lobes of the eight have been folded together in the back of the page to create the usual image of a Möbius ladder.

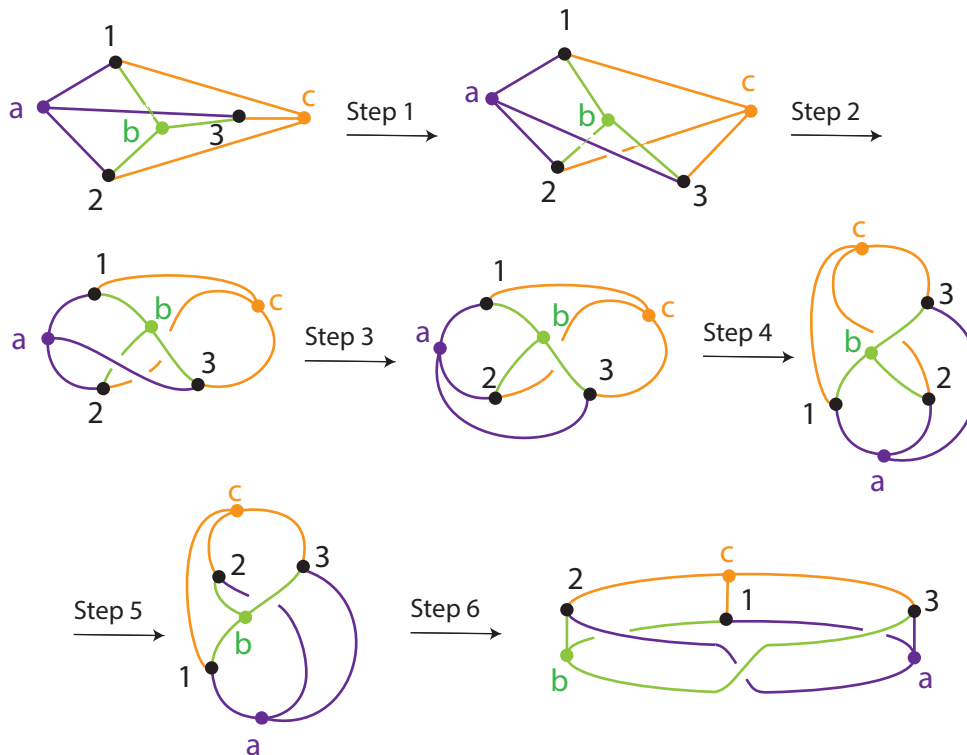


FIGURE 25. A deformation of the  $K_{3,3}$  in the M-cluster of nitrogene to a Möbius ladder with three rungs.

Since a Möbius ladder with three rungs is a conformation of the non-planar graph  $K_{3,3}$  in space, any protein structure containing such a Möbius ladder is necessarily non-planar. However, just like knowing which particular knot is contained in a protein structure gives us more information about the structure than knowing that it contains an unspecified knot, knowing that a protein contains a Möbius ladder gives us more information about the structure than knowing that it contains an unspecified conformation of the abstract graph  $K_{3,3}$ .

In fact, Möbius ladders can have any number of rungs, and as long as the number of rungs is at least three the Möbius ladder will necessarily be non-planar. Figure 26 depicts an iron-sulfur cluster connected to the backbone of a protein structure which we show as follows contains a Möbius ladder with four rungs. We begin by illustrating the underlying topological form of the structure on the right of Figure 26.

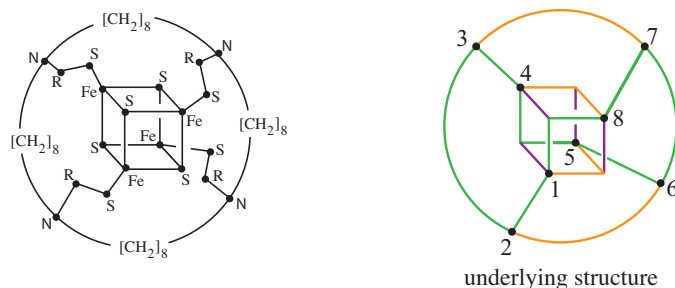


FIGURE 26. The underlying topological structure of a protein structure with the  $\text{Fe}_4\text{S}_4$  cluster .

In Figure 27 we omit the four purple edges from the cube-like form in the center and then deform the remaining structure to a Möbius ladder with four rungs.

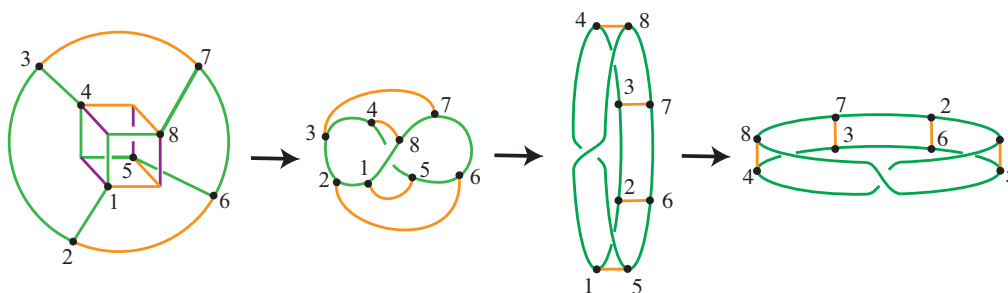


FIGURE 27. A deformation taking a protein structure with an iron-sulfur cluster to a Möbius ladder with four rungs.

Möbius ladders can also be found in protein structures that do not contain metal clusters. For example, one class of polypeptides that can easily be shown to contain Möbius ladders is the family of cyclotides. These small cyclized peptides (typically about 30 amino acids in length) contain many disulfide bonds, and because of their head-to-tail cyclization, we can represent them with a circular backbone. Figure 28 illustrates the highly conserved disulfide connectivity of a cyclotide, along with its deformation into a Möbius ladder with three rungs. In order to obtain the Möbius ladder configuration, first edge connecting vertices 1 and 4 is lifted over and to the right of the structure. Then, the two black lobes are rounded out such that the top lobe contains vertices 1, 2, and 3, while the bottom lobe contains vertices 4, 5, and 6. The two lobes are then folded together towards the back of the page to create the familiar image of the Möbius ladder. The image in the top left of Figure 28 depicting a typical cyclotide with three disulfide bonds is often described as a “cysteine knot motif.” This description seems somewhat inaccurate, since the cyclotide does not actually contain a knot. However, we saw above that the cyclotide can be deformed to a Möbius ladder with three rungs, and hence we can interpret a “cysteine knot motif” to mean that the cyclotide has a topologically non-planar underlying structure. Because this connectivity is

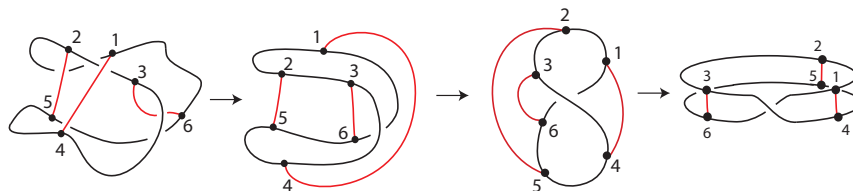


FIGURE 28. This deformation illustrates that the cyclotide “knot motif” is a Möbius ladder with three rungs.

highly preserved among cyclotides, we can generalize this non-planarity to nearly all cyclotides that contain this connectivity.

Note that cyclotides have been grouped into two structural categories: Möbius cyclotides and bracelet cyclotides. Much less common than the bracelet structure, Möbius cyclotides contain a cis-proline induced backbone twist, making their backbones resemble the edge of a Möbius strip (and hence their name makes sense). However, in fact both the bracelet cyclotides and the Möbius cyclotides contain Möbius ladders with three rungs as illustrated in Figure 28.

Even for proteins that are not cyclized we can identify knots, links and non-planar spatial graphs by using a variation on the topological models to trap knots in open linear protein structures that were introduced by Millett and others [11, 16, 18, 20, 21] (see Section 1). In particular, recall that one approach to modeling a knotted linear protein structure is to place it in the center of a very large ball, and then extend the ends of the chains out away from the feature of interest and connect them by an arc within the boundary of the ball. We can use a variation on this method for the green protein to identify the structure depicted on the left in Figure 29 as linked. The purple protein has been cyclized as a result of a disulfide bond, shown in red. By joining the termini of the green linear protein, we obtain a diagram of a link.

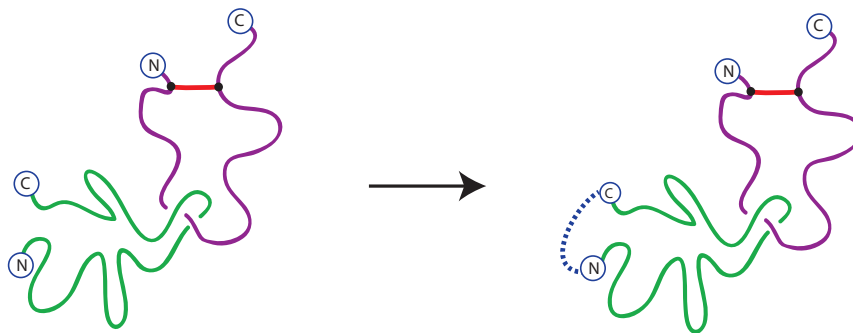


FIGURE 29. Connecting the termini of the green protein allows us to visualize this structure as a link.

The idea of connecting the N- and C- termini by an arc that is far from any features of interest can also enable us to use Möbius ladders to prove non-planarity of the underlying structures of certain proteins. For example, in Figure 30, we show that the nerve growth factor together with three disulfide bonds can be deformed into a Möbius ladder with three rungs. The first step of the deformation is to deform the edge between vertices 3 and 6 by lifting it over the top arc of the backbone so that it moves from being in front of the backbone to being behind the backbone, and to move the edge between vertices 2 and 5 down to the bottom of the diagram. The second step is to shorten the edges between vertices 1 and 4 and between vertices 3 and 6 causing the backbone to cross over itself. The third step is to reshape the twisted circle. Finally, in the fourth step we move vertex 2 up and to the right as we shape the twisted circle into the standard form of a Möbius ladder. Like the cyclotide, this protein's disulfide connectivity is also referred to as a “cysteine knot motif”, which again can be interpreted as referring to its non-planarity.

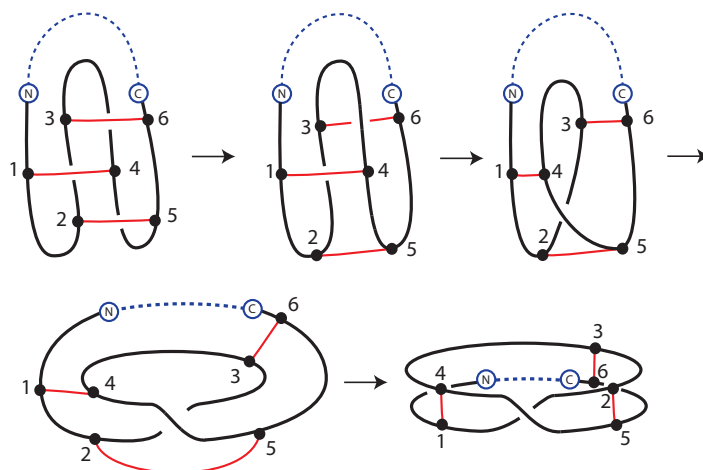


FIGURE 30. A deformation of the underlying topological structure of the nerve growth factor to a Möbius ladder with three rungs.

The middle steps of the deformations illustrated in Figures 25, 27, 28, and 30 each include a conformation consisting of a deformed circle with three or four red edges which cross over one another. It might be tempting to assume that any such conformation can be deformed into a Möbius ladder. But this is not always the case. To illustrate this, consider the connectivity of chymotrypsin inhibitor from Taiwan Cobra, shown in Figure 31. Observe that in this case we can deform the structure so that it lies in the plane of the paper. In particular, this means that in contrast with Figures 25, 27, 28, and 30, the structure in Figure 31 does not contain a Möbius ladder.

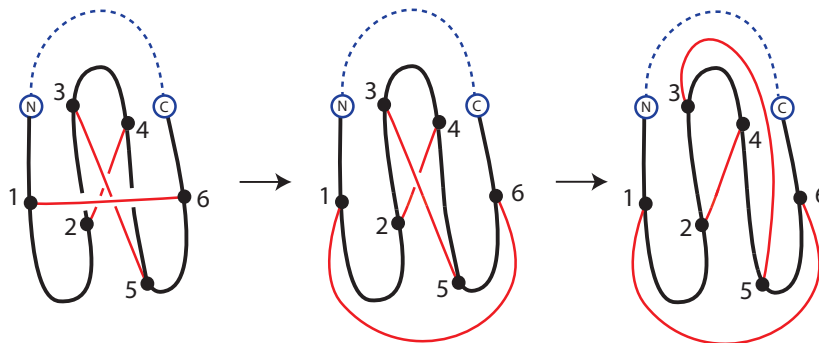


FIGURE 31. A deformation illustrating that not all proteins with three disulfide bonds contain Möbius ladders. This one, as shown, can be deformed in to a plane.

### 3. Hypothetical Structures

In this section, we propose hypothetical structures illustrating how certain non-planar protein structures might occur. First we consider the  $5_1$  knot, which has not yet been identified in a protein structure. Starting with a protein whose backbone is in the form of a trefoil knot, Figure 32 illustrates how an additional threading of one end of the backbone through the loop of the trefoil could result in a  $5_1$  knot. Note that the existence of many protein trefoil knots makes this realization of the  $5_1$  knot seem plausible, despite Taylor's argument that the  $5_1$  knot is an unlikely motif [32].

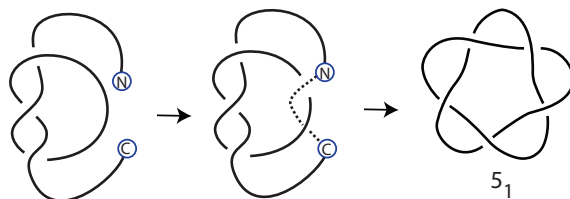
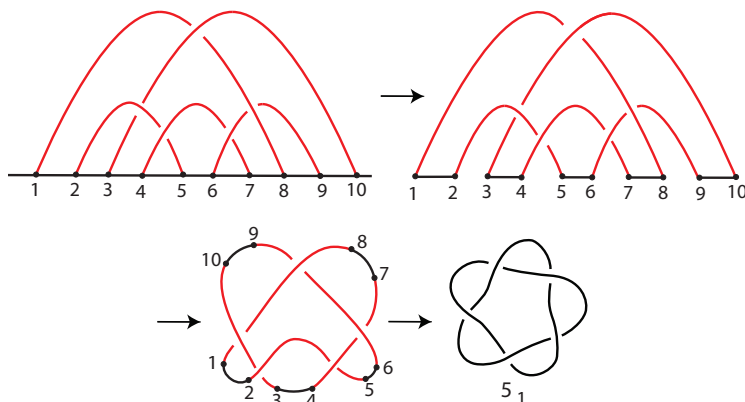


FIGURE 32. Starting with the trefoil knot, only one threading is required to obtain the  $5_1$  knot.

Another approach to obtaining a  $5_1$  knot in a protein structure is to make use of disulfide bonds. In particular, consider a hypothetical protein structure whose polypeptide backbone has five disulfide bonds with connectivity 1-6, 2-3, 4-5 and crossings as illustrated in the image on the left in Figure 33. If we ignore every other segment on the backbone, we obtain the second figure, which can then be deformed as illustrated to get the  $5_1$  knot.

FIGURE 33. A connectivity that could contain the  $5_1$  knot.

Along these same lines, in Figure 34 we show that a key ring link with three components can hypothetically be found in a single protein which has three disulfide bonds. In particular, consider a hypothetical protein with six cysteine residues on the backbone with disulfide connectivity 1-6, 2-3, 4-5 and crossings as illustrated. In this case (as opposed to that of the  $5_1$  knot), we need to join the ends of the backbone into a circle as we did in Figure 29 in order to close the blue ring. Then by omitting the black edges of the circle we obtain the three component key ring link as illustrated on the right. Note that this key ring link is different from the one illustrated in Figure 12 because it uses disulfide bonds rather than cofactors.

A similar construction can be used to create a hypothetical protein structure containing the Whitehead link, a motif that has yet to be identified in any protein structure. This link consists of a circular loop that passes through both lobes of a twisted circle, as shown on the right of Figure 35. Such a structure would be particularly noteworthy because the two rings are linked yet have linking number of zero. On the left, we show how this two-component link could hypothetically be obtained from a protein backbone with three disulfide bonds having 1-6, 2-4, 3-5 disulfide connectivity and crossings as shown. As with the structure in Figure 34, we need to join the ends of the backbone into a circle in order to close one of the rings. Again by omitting the black edges of the circle we obtain the desired link.

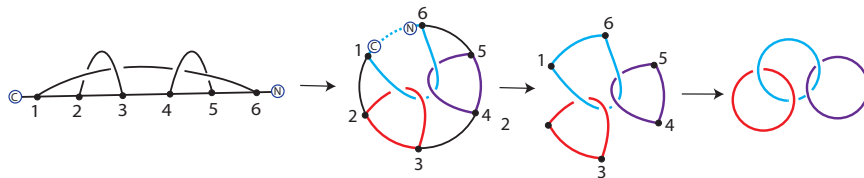


FIGURE 34. Scheme showing a possible connectivity that would contain a three component key ring link

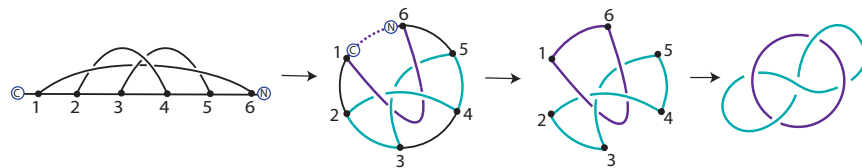


FIGURE 35. Scheme showing a possible disulfide connectivity that would contain the Whitehead Link.



## CHAPTER 4

### Conclusion

Topologically complex proteins can offer valuable perspectives for understanding protein stability, folding mechanisms and degradation pathways. We have shown that such complexity can be found in protein structures in various ways as the result of knotting, linking, or non-planar graphs. Not only can the polypeptide backbone of a protein structure be knotted or a pair of backbones be linked, but knots, links, and non-planar graphs can be found as the result of cofactors and disulfide bonds.

Only four topologically distinct knots have been identified in proteins thus far, and the links that have been identified in proteins all have relatively simple linking (i.e. with linking numbers equal to  $\pm 1$ ). However, as we saw in our hypothetical structures in Section 3, there might be many more complex knots and links identified in proteins if the connectivity of disulfide bonds is considered along with protein backbones.

Finally, we have seen that proteins containing metal structures can include non-planar motifs other than knots or links. In particular, nitrogenase contains several non-planar  $K_{3,3}$  and  $K_5$  graphs. Furthermore, Möbius ladders (i.e., ladders in the form of a Möbius strip) can be found in proteins with metal structures including the M-cluster of nitrogenase, as well as in proteins without metal structures including cyclotides. This opens the question of what other interesting spatial graphs might be found in protein structures if cofactors and cysteine disulfide bonds are included in the topological analysis.



## CHAPTER 5

### Acknowledgements

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